

Revised Structure of the Triterpene Polygalacic Acid: Configuration of the C-16 Hydroxy-group

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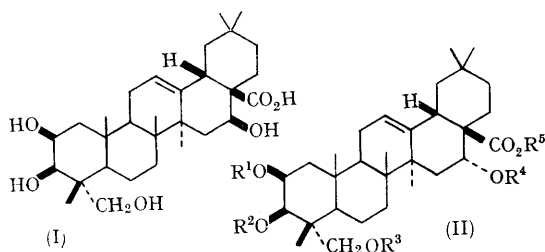
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POLYGALACIC acid is the triterpenoid tetrahydroxy-acid isolated from *Polygala paenea* L.¹ and its structure has been assigned as 2 β ,3 β ,16 β ,23-tetrahydroxyolean-12-en-28-oic acid (I).² In our study on saponins of the roots of *Platycodon grandiflorum*, we isolated polygalacic acid,† together with platycodigenin [m.p. 250—252°, [α]_D +46.5° (EtOH)] which is an unidentified penta-hydroxy-acid C₃₀H₄₈O₇, isolated previously³ as a sapogenin of this plant. We report evidence which suggests that the structure of polygalacic acid should be revised to 2 β ,3 β ,16 α ,23-tetrahydroxyolean-12-en-28-oic acid (IIa).

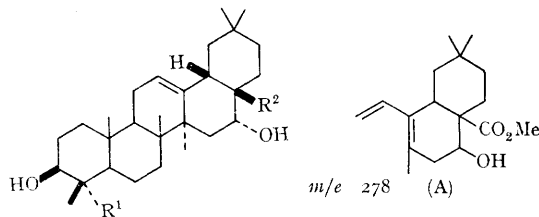
Polygalacic acid, C₃₀H₄₈O₆, m.p. 295—300°, [α]_D + 47.1° (pyridine) and its methyl ester, m.p. 235—241°, [α]_D + 44.5° (EtOH), were identified with authentic samples.‡ Acetylation of the methyl ester (IIb) with acetic anhydride and pyridine at room temperature followed by t.l.c. separation on silica gel G (toluene-ethyl acetate, 3:1) gave, in addition to the tetra-acetate (IIc),² m.p. 169—174°, an amorphous triacetate (IIId). The mass spectrum of (IIId) (*M*⁺ 644) shows *m/e*

278, attributable to the retro-Diels-Alder fragment⁴ (A), as observed in the spectrum of the methyl ester (IIb), but no peak of *m/e* 320 for an acetoxy-group. This fact reveals that, of the four hydroxy-groups in polygalacic acid, the one located at C(16) is the most difficult to acetylate and therefore it may be axial 16 α -hydroxyl. In the n.m.r. spectra (60 Mc./sec., CDCl₃), a signal at τ 4.33 (poorly resolved triplet, *W*₁ 8 c./sec.)² observed for the tetra-acetate (IIc) shifts to τ 5.47 for the tri-acetate (IIId). These chemical shifts and signal shapes are in good agreement with those assigned for the 16 β -H in the di- and mono-acetates of quillaic acid methyl ester (III)⁵ which has an axial 16 α -hydroxy-group.

These results were supported by the following experiment. Treatment of the methyl ester (IIb) with acetone and toluene-*p*-sulphonic acid gave the 2 β ,3 β -acetone (IIe),² m.p. 204—206°, [α]_D +66.4° (CHCl₃), and the amorphous 3 β ,23-acetone in a 3:2 ratio. The 2 β ,3 β -acetone (IIe) on acetylation with acetic anhydride and pyridine on a steam bath afforded the acetone-diacetate



	R ¹	R ²	R ³	R ⁴	R ⁵
a;	H	H	H	H	H
b;	H	H	H	H	Me
c;	Ac	Ac	Ac	Ac	Me
d;	Ac	Ac	Ac	H	Me
e;		Me ₂ C	H	H	Me
f;		Me ₂ C	Ac	Ac	Me
g;	H	H	Ac	Ac	Me
h;	H	Ac	Ac	Ac	Me
i;	MeSO ₂	Ac	Ac	Ac	Me



(III; R¹=CHO, R²=CO₂Me)

(IV; R¹=R²=CH₂OH)

† O. Tanaka *et al.* have recently reported an independent isolation of polygalacic acid from the same source. (See abstract of the 88th Annual Meeting of the Pharmaceutical Society of Japan, April 1968, p. 216.)

‡ We thank Dr. Rondest for these samples.

(II_f), m.p. 121—124° (from MeOH), $[\alpha]_D + 30.6^\circ$ (CHCl₃), which on hydrolysis with 70% aqueous acetic acid yielded the amorphous 16,23-diacetate (II_g). Partial acetylation of (II_g) with acetic anhydride–pyridine–chloroform (1:5:25) at room temperature for 40 hr. gave predominantly the 3 β ,16 α ,23-triacetate (II_h), M^+ 644, [n.m.r. (CDCl₃) τ 7.95, 7.92, and 7.89 (3H each of acetyls, s), 6.38 (3H of CO₂Me, s), 6.25 (2H at C-23, bs), 5.78 (1H at C-2, m), 5.10 (1H at C-3, d, J 3.7 c./sec.), 4.57 (1H at C-12, bt), and 4.35 (1H at C-16, bt). Treatment of (II_h) with methanesulphonyl chloride in pyridine afforded the triacetyl-methanesulphonate (II_i) [ν_{\max} (CHCl₃) 1345, 1173, and 912 cm.⁻¹; no

OH]. Reduction of (II_i) with lithium aluminium hydride in boiling tetrahydrofuran yielded a tetraol, C₃₀H₅₀O₄, m.p. 256—260°, $[\alpha]_D + 47.6^\circ$ (CHCl₃), which was identified with authentic olean-12-ene-3 β ,16 α ,23,28-tetraol (IV) derived from lithium aluminium hydride reduction of quillaic acid methyl ester (III).

From the above chemical correlation, the structure of polygalacic acid has been shown to be (II_a), in which it has the 16 α - not 16 β -hydroxy-configuration, as previously suggested from circumstantial evidence.²

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² J. Rondet and J. Polonsky, *Bull. Soc. chim. France*, 1963, 1253.

³ M. Tsujimoto, *J. Agric. Chem. Soc. Japan*, 1940, **16**, 613 and preceding papers.

⁴ H. Budzikiewicz, J. M. Wilson, and C. Djerassi, *J. Amer. Chem. Soc.*, 1963, **85**, 3688.

⁵ L. Ruzicka, B. Bischof, E. C. Taylor, A. Meyer, and O. Jeger, *Coll. Czech. Chem. Comm.*, 1950, **15**, 893.